

## The *exo* and *endo* isomers of [*N,N'*-bis(salicylidene)-1,2-diphenyl-(*RS,SR*)-1,2-ethanediyl]diaminato]oxo-vanadium(IV)

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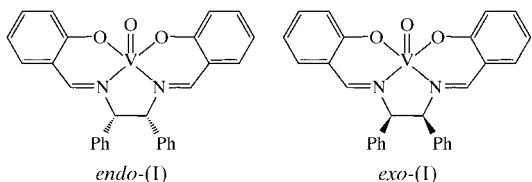
Data validation number: IUC0000085

The title complexes, *exo*- and *endo*-[VO(C<sub>28</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>)], show monomeric structures with a distorted square-pyramidal coordination. The two phenyl groups on the five-membered *N,N'*-chelate ring are both on the same side as the oxo ligand for the *exo* isomer and on opposite sides for the *endo* isomer.

### Comment

The crystal structures of the monomeric and polymeric forms of [VO{sal-(*RR*)-stien}] were reported by Nakajima *et al.* (1996). The crystal structures and the thermal isomerization of [VO(3-EtOsal-meso-stien)] have been investigated by the authors (Hoshina *et al.*, 1999), together with the thermal dehydrogenation in the solid state (Hoshina *et al.*, 1998, 2000).

The title complex, [VO(sal-meso-stien)], (I), did not show thermal isomerization in the solid state below 463 K. The *endo* isomer was more reactive in dehydrogenation upon heating in air than the *exo* isomer.



### Experimental

The *exo*-[VO(sal-meso-stien)] isomer was prepared selectively by the reaction of vanadium(IV) oxide sulfate (0.25 g, 1 mmol) with the Schiff base ligand H<sub>2</sub>sal-meso-stien (0.42 g, 1 mmol) in methanol containing pyridine (yield 74%). The *endo* isomer was prepared separately by the reaction of vanadium(IV) oxide sulfate (0.25 g, 1 mmol) and H<sub>2</sub>sal-meso-stien (0.42 g, 1 mmol) in methanol without pyridine (yield 86%). The *endo*-[VO(sal-meso-stien)] isomer was

eluted more slowly than the *exo* isomer by high-performance liquid reverse-phase chromatography using CH<sub>3</sub>CN/H<sub>2</sub>O (1:1 v/v) as eluent.

### The *exo* isomer

#### Crystal data

[VO(C<sub>28</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>)]  
*M*<sub>r</sub> = 485.43  
Monoclinic, *P*2<sub>1</sub>/*n*  
*a* = 13.278 (2) Å  
*b* = 11.340 (3) Å  
*c* = 15.940 (2) Å  
 $\beta$  = 101.12 (1) $^\circ$   
*V* = 2355.1 (7) Å<sup>3</sup>  
*Z* = 4

*D*<sub>x</sub> = 1.369 Mg m<sup>-3</sup>  
Mo *K* $\alpha$  radiation  
Cell parameters from 25 reflections  
 $\theta$  = 14.9–15.0 $^\circ$   
 $\mu$  = 0.454 mm<sup>-1</sup>  
*T* = 297 (1) K  
Prismatic, green  
0.7 × 0.2 × 0.2 mm

#### Data collection

Rigaku AFC-7R diffractometer  
 $\theta$ –2*θ* scans  
Absorption correction: by integration (Coppens *et al.*, 1965)  
*T*<sub>min</sub> = 0.881, *T*<sub>max</sub> = 0.915  
5646 measured reflections  
5417 independent reflections  
3380 reflections with *I* > 2*σ*(*I*)

*R*<sub>int</sub> = 0.016  
 $\theta$ <sub>max</sub> = 27.5 $^\circ$   
*h* = -17 → 0  
*k* = 0 → 14  
*l* = -20 → 20  
3 standard reflections every 150 reflections  
intensity decay: none

#### Refinement

Refinement on *F*<sup>2</sup>  
*R*(*F*) = 0.047  
*wR*(*F*<sup>2</sup>) = 0.133  
*S* = 1.03  
5417 reflections  
307 parameters  
H-atom parameters not refined

*w* = 1/[ $\sigma$ <sup>2</sup>(*F*<sub>o</sub><sup>2</sup>) + (0.0505*P*)<sup>2</sup> + 1.8247*P*]  
where *P* = (*F*<sub>o</sub><sup>2</sup> + 2*F*<sub>c</sub><sup>2</sup>)/3  
( $\Delta/\sigma$ )<sub>max</sub> = 0.001  
 $\Delta\rho_{\text{max}}$  = 0.81 e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}}$  = -0.33 e Å<sup>-3</sup>

### The *endo* isomer

#### Crystal data

[VO(C<sub>28</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>)]  
*M*<sub>r</sub> = 485.43  
Orthorhombic, *P*bca  
*a* = 17.671 (6) Å  
*b* = 21.635 (7) Å  
*c* = 12.565 (4) Å  
*V* = 4803 (2) Å<sup>3</sup>  
*Z* = 8  
*D*<sub>x</sub> = 1.342 Mg m<sup>-3</sup>

Mo *K* $\alpha$  radiation  
Cell parameters from 25 reflections  
 $\theta$  = 13.4–14.9 $^\circ$   
 $\mu$  = 0.445 mm<sup>-1</sup>  
*T* = 297 (1) K  
Prismatic, green  
0.7 × 0.5 × 0.2 mm

#### Data collection

Rigaku AFC-7R diffractometer  
 $\theta$ –2*θ* scans  
Absorption correction: by integration (Coppens *et al.*, 1965)  
*T*<sub>min</sub> = 0.787, *T*<sub>max</sub> = 0.912  
5807 measured reflections  
5516 independent reflections  
4206 reflections with *I* > 2*σ*(*I*)

*R*<sub>int</sub> = 0.024  
 $\theta$ <sub>max</sub> = 27.5 $^\circ$   
*h* = -8 → 22  
*k* = -10 → 28  
*l* = -6 → 16  
3 standard reflections every 150 reflections  
intensity decay: none

#### Refinement

Refinement on *F*<sup>2</sup>  
*R*(*F*) = 0.055  
*wR*(*F*<sup>2</sup>) = 0.193  
*S* = 1.04  
5516 reflections  
307 parameters  
H-atom parameters not refined

*w* = 1/[ $\sigma$ <sup>2</sup>(*F*<sub>o</sub><sup>2</sup>) + (0.0776*P*)<sup>2</sup> + 11.0899*P*]  
where *P* = (*F*<sub>o</sub><sup>2</sup> + 2*F*<sub>c</sub><sup>2</sup>)/3  
( $\Delta/\sigma$ )<sub>max</sub> = 0.0001  
 $\Delta\rho_{\text{max}}$  = 0.75 e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}}$  = -0.83 e Å<sup>-3</sup>

All H-atom positions were calculated geometrically and fixed with *U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(parent atom).

For both compounds, data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1993); cell

refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1999); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); software used to prepare material for publication: *TEXSAN*.

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